

Structures and Biomimetic Synthesis of Novel α -Pyrone Polyketides of an Endophytic *Penicillium* sp. in *Catharanthus roseus*

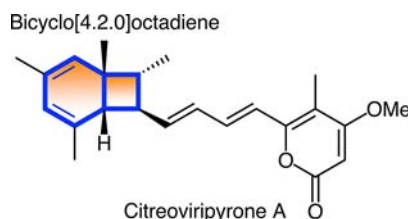
Teigo Asai,^{*,†} Dan Luo,[†] Kouwa Yamashita,[‡] and Yoshiteru Oshima^{*,†}

Graduate School of Pharmaceutical Sciences, Tohoku University, Aoba-yama, Aoba-ku, Sendai 980-8578, Japan, and Tohoku Pharmaceutical University, 4-4-1 Komatsushima, Aoba-ku, Sendai 981-8558, Japan

tasai@m.tohoku.ac.jp

Received December 21, 2012

ABSTRACT



Novel polyketides, citreoviripyrone A (1) and B (2), known citreomontanin (3), and (–)-citreoviridin (4) were isolated from the mycelium of the endophytic fungus. The endophytic fungus, which belongs to the genus *Penicillium*, was separated from surface-sterilized healthy leaves of *Catharanthus roseus*. The structures of 1 and 2 were determined on the basis of NMR data, and 1 was characterized as an α -pyrone polyketide featuring bicyclo[4.2.0]octadiene. The biomimetic synthesis of 1 and 2 from 3 elucidated a plausible biosynthetic pathway. Both Zn(II)-type and NAD⁺-dependent histone deacetylase inhibitors significantly enhanced the production of 1 and 3.

Bicyclo[4.2.0]octadiene is crucial for structurally and biologically interesting natural products. Several natural products featuring the bicyclo skeleton have been reported from various sources, including the Lauraceae plant

(endiandric acid D and E),¹ saccoglossan mollusc (ocellapyrone A),² *Streptomyces* (SNF4435 C and D),³ and marine-derived fungus (shimalactone A and B) (Figure 1).⁴

These natural products may arise from the 8π – 6π electrocyclization cascade route from their corresponding polyene precursors, most of which are highly methylated.⁵

Several *Penicillium* fungi, such as *P. citreoviride* isolated from “yellowed rice”,⁶ produce (–)-citreoviridin (4)⁷ and citreomontanin (3).⁸ 4 is a well-known mycotoxin composed of three parts: α -pyrone, polyene, and a highly oxidized tetrahydrofuran ring (Figure 2). Due to the structural similarities, 3 is thought to be biosynthetically related to 4. Although 3 should biosynthetically form compounds with a bicyclo[4.2.0]octadiene in the structure,

[†] Tohoku University.

[‡] Tohoku Pharmaceutical University.

(1) Banfield, J. E.; Black, D. St. C.; Johns, S. R.; Willing, R. I. *Aust. J. Chem.* **1982**, *35*, 2247–2256.

(2) Manzo, E.; Ciavatta, M. L.; Gavagnin, M.; Mollo, E.; Wahidulla, S.; Cimino, G. *Tetrahedron Lett.* **2005**, *46*, 465–468.

(3) (a) Kurosawa, K.; Takahashi, K.; Tsuda, E. *J. Antibiotic* **2001**, *54*, 541–547. (b) Takahashi, K.; Tsuda, E.; Kurosawa, K. *J. Antibiotic* **2001**, *54*, 548–553.

(4) (a) Wei, H.; Itoh, T.; Kinoshita, M.; Kotoku, N.; Aoki, S.; Kobayashi, M. *Tetrahedron* **2005**, *61*, 8054–8058. (b) Wei, H.; Itoh, T.; Kinoshita, M.; Kotoku, N.; Aoki, S.; Kobayashi, M. *Heterocycles* **2006**, *68*, 111–123.

(5) (a) Miller, A. K.; Trauner, D. *Angew. Chem., Int. Ed.* **2005**, *44*, 4602–4606. (b) Beaudry, C. M.; Trauner, D. *Org. Lett.* **2002**, *4*, 2221–2224. (c) Moses, J. E.; Baldwin, J. E.; Marquez, R.; Adlington, R. M.; Cowley, A. R. *Org. Lett.* **2002**, *4*, 3731–3734. (d) Jacobsen, M. F.; Moses, J. E.; Adlington, R. M.; Baldwin, J. E. *Org. Lett.* **2005**, *7*, 641–644. (e) Müller, M.; Kusebauch, B.; Liang, G.; Beaudry, C. M.; Trauner, D.; Hertweck, C. *Angew. Chem., Int. Ed.* **2006**, *45*, 7835–7838. (f) Eade, S. J.; Walter, M. W.; Byrne, C.; Odell, B.; Rodriguez, R.; Baldwin, J. E.; Adlington, R. M.; Moses, J. E. *J. Org. Chem.* **2008**, *73*, 4830–4839.

(6) Sakabe, N.; Goto, T.; Hirata, Y. *Tetrahedron* **1977**, *33*, 3077–3081.

(7) (a) Nagel, D. W.; Steyn, P. S.; Scott, D. B. *Phytochemistry* **1972**, *11*, 627–630. (b) Cole, R. J.; Dorner, J. W.; Cox, R. H.; Hill, R. A.; Cluter, H. G.; Wells, J. M. *Appl. Environ. Microbiol.* **1981**, *42*, 677–681.

(8) Rebuffat, S.; Davoust, D.; Molho, L.; Molho, D. *Phytochemistry* **1980**, *19*, 424–431.

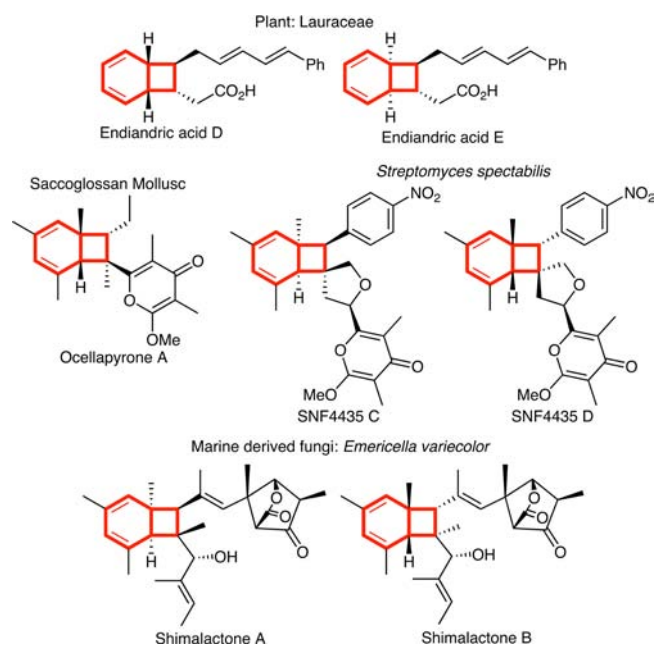


Figure 1. Previously reported compounds possessing a bicyclo[4.2.0]octadiene core (red bold line).

these types of compounds have yet to be isolated from *Penicillium* fungi.

In our search for novel natural products of endophytic fungi, which are a rich source of structurally and biologically interesting compounds,⁹ we have separated an endophytic *Penicillium* sp. CR07 from surface-sterilized healthy leaves of *Catharanthus roseus*, which is a well-known producer of a vinca alkaloid. A chemical study of MeOH extracts of the mycelium of the *Penicillium* sp. CR07 afforded novel α -pyrone polyketides, citreoviripyrene A (**1**) and citreoviripyrene B (**2**), along with **3** and **4**. Herein we report the structures of **1** and **2**, their biomimetic conversion from **3**, and their cell growth inhibitory activity. In addition, we discuss the effects of both Zn(II)-type and NAD⁺-dependent HDAC inhibitors on the polyketide production in the *Penicillium* fungus.

Endophytic *Penicillium* sp. CR07 was cultivated in a potato dextrose broth (PDB) for 21 days at 25 °C. The mycelium (155.6 g) was extracted twice with MeOH, and the MeOH extract (33.9 g) was partitioned between EtOAc and H₂O. The EtOAc extract (10.8 g) was separated by silica gel column chromatography and preparative TLC to afford citreoviripyrene A (**1**, 19.1 mg), B (**2**, 1.2 mg), citreomontanin (**3**, 35.0 mg), and (–)-citreoviridin (**4**, 1.9 g).

The HRFABMS of citreoviripyrene A (**1**) at m/z 353.2133 [$M + H$]⁺ (calcd: m/z 353.2117) yielded a molecular formula of C₂₃H₂₈O₃, which requires ten degrees of unsaturation. The UV absorption at 354 nm (log ϵ = 4.33) suggested that the molecule contained an extended

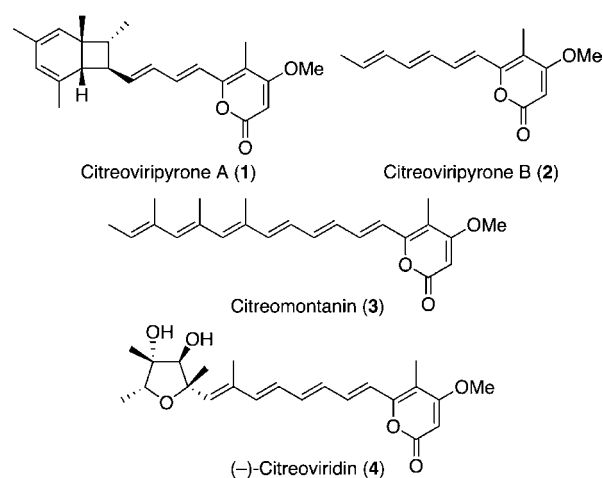


Figure 2. Structures of **1**–**4**.

conjugated system. The ¹³C NMR and DEPT spectra implied the presence of one ester carbonyl, five quaternary sp² carbons, seven tertiary sp² carbons, one quaternary carbon, three methines, one methoxy methyl, and five methyls (Table 1). Comparing the ¹H and ¹³C NMR spectra of **1** to those of **4** indicated the presence of a 3,4,5-trisubstituted 3-methoxy- α -pyrone moiety [δ_C 163.7 (C-1), 88.5 (C-2), 170.7 (C-3), 107.2 (C-4), 154.6 (C-5), and 56.1 (OMe); δ_H 5.48 (s, H-2), 1.94 (s, H₃-22), and 3.82 (s, OMe) (Tables 1 and S1)], which was consistent with the HMBC correlations of H-2/C-1, C-3, C-4; H₃-22/C-3, C-4, C-5; and OMe/C-3 (Figure 3). The ¹H–¹H COSY spectrum exhibited sequential correlations of H-6–H-7–H-8–H-9–H-10–H-17–H-18 and H-10–H-11 (Figure 3). The HMBC correlations of H₃-19/C-11, C-15, C-16, C-17; H₃-20/C-13, C-14, C-15; and H₃-21/C-11, C-12, C-13 indicated a pentasubstituted bicyclo[4.2.0]octadiene structure (C-10–C-21) (Figure 3). Additionally, the HMBC correlations of H-6/C-4 and H-6/C-5 determined the C-5/C-6 linkage. The large coupling constants confirmed that the geometries of the C-6/C-7 and C-8/C-9 double bonds were both *E* ($J_{H-6/H-7}$ = 15.0 Hz and $J_{H-8/H-9}$ = 15.1 Hz).

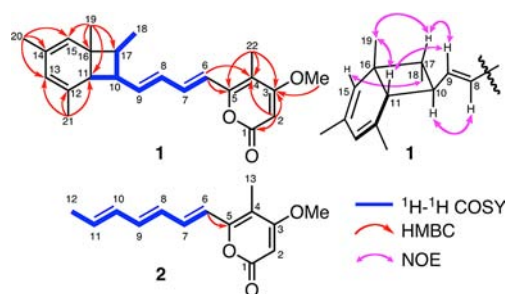


Figure 3. Selected HMBC, ¹H–¹H COSY, and NOE correlations of **1** and **2**.

(9) (a) Guunatilaka, A. A. L. *J. Nat. Prod.* **2006**, 69, 509–526. (b) Guo, B.; Wang, Y.; Sun, X.; Tang, K. *Appl. Biochem. Microbiol.* **2008**, 44, 136–142.

Table 1. ^{13}C (150 MHz) and ^1H NMR (600 MHz) Data for **1** and **2**^a

	1		2	
	^{13}C	^1H (multi, J in Hz)	^{13}C	^1H (multi, J in Hz)
1	163.7 C	—	163.7 C	—
2	88.5 CH	5.48 (s)	88.5 CH	5.50 (s)
3	170.7 C	—	170.6 C	—
4	107.2 C	—	107.4 C	—
5	154.6 C	—	154.6 C	—
6	117.7 CH	6.26 (d, 15.0)	118.3 CH	6.30 (d, 15.0)
7	136.4 CH	7.13 (dd, 15.0, 10.7)	136.2 CH	7.18 (dd, 15.0, 11.0)
8	128.0 CH	6.15 (dd, 15.1, 10.7)	129.1 CH	6.25 (dd, 15.0, 11.0)
9	144.0 CH	6.05 (dd, 15.1, 8.8)	138.5 CH	6.46 (dd, 15.0, 10.6)
10	51.8 CH	2.46 (q, 8.8)	131.5 CH	6.15 (dd, 15.0, 10.6)
11	49.8 CH	2.19 (d, 8.8)	133.6 CH ₃	5.90 (m)
12	135.1 C	—	18.5 CH ₃	1.83 (d, 6.6)
13	120.9 CH	5.37 (brs)	8.8 CH ₃	1.96 (s)
14	129.9 C	—	—	—
15	122.7 C	4.97 (brs)	—	—
16	40.4 C	—	—	—
17	51.2 CH	2.15 (m)	—	—
18	13.2 CH ₃	0.99 (d, 8.8)	—	—
19	28.3 CH ₃	1.03 (s)	—	—
20	22.3 CH ₃	1.75 (brs)	—	—
21	22.0 CH ₃	1.66 (brs)	—	—
22	8.8 CH ₃	1.94 (s)	—	—
OMe	56.1 CH ₃	3.82 (s)	56.1 CH ₃	3.82 (s)

^a Recorded in CDCl_3 . Assignments for all compounds were based on DQF-COSY, HMQC, and HMBC experiments.

1D NOE experiments determined the relative stereochemistries around the bicyclo moiety. The NOEs of H-9/H-11, H-17 and H₃-19/H-11, H-17 indicated that the H-11, H-17, H₃-19, and C-9 were situated on the same side. In addition, the NOE of H₃-18/H-15 supported the *anti* relationship between H₃-18 and H₃-19. Citreoviripyrone A (**1**) should be racemic due to the minimal optical rotation of **1** ($[\alpha]_{\text{D}}^{25} = \pm 0.00$ ($c = 0.16$, CHCl_3)) and the fact that the CD spectrum did not display a remarkable Cotton effect.

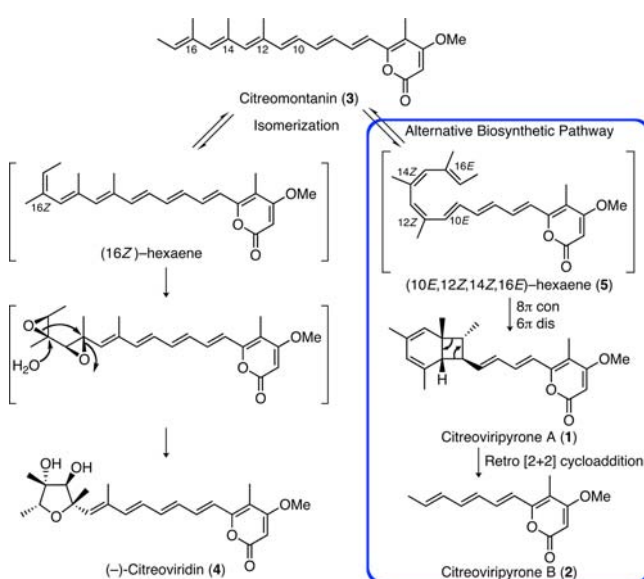
The molecular formula of citreoviripyrone B (**2**) was $\text{C}_{14}\text{H}_{16}\text{O}_3$ [HREIMS: m/z 232.1084 [M]⁺ (calcd: m/z 232.1099)], which requires seven degrees of unsaturation. Similar to compound **1**, the UV absorption at 356 nm ($\log \varepsilon = 4.07$) showed the presence of an extended conjugated system. The ^{13}C NMR and DEPT spectra indicated the presence of one ester carbonyl, three quaternary sp^2 carbons, seven tertiary sp^2 carbons, one methoxy methyl, and two methyls (Table 1). Comparing the ^{13}C NMR spectrum of **2** to that of **1** revealed the presence of a 3,4,5-trisubstituted 3-methoxy- α -pyrone moiety (C-1–C-5, C-13, OMe) (Table 1). Sequential ^1H – ^1H COSY correlations (H-6–H₃-12) and HMBC correlations of H-6/C-5 confirmed the triene (C-6–C-12) was linked to C-5 on the α -pyrone (Figure 3). The large coupling constants

of H-6/H-7, H-8/H-9, and H-10/H-11 (15.0 Hz for all three J -values) indicated that the three olefins in the triene had *trans* geometries (Table 1). Thus the structure of citreoviripyrone B (**2**) was determined to be as shown Figure 2.

Compounds **1** and **3** have common structural features, including the same molecular formula. Both have the same number of methyl groups at the same substituted positions. Additionally, both have a trisubstituted α -pyrone. Considering these structural similarities, **3** is likely a putative biogenetic precursor of **1** (Scheme 1).

Treatment of all-*(E)*-hexaene **3** with dichlorobis(acetonitrile)palladium(II) [$\text{PdCl}_2(\text{MeCN})_2$], which is a well-known reagent to isomerize a conjugated double-bond system,^{5c} in DMF at 70 °C for 1 h, afforded **1** in 13% yield with a trace of **2**. In contrast, **2** was isolated in 18% yield as the major product for the same reaction after 14 h (Scheme S1). TLC monitoring (0–14 h) of the reaction showed that **1** was generated initially but was gradually converted into **2**. Presumably, **3** yielded **1** through intermediate **5** by a thermal 8π conrotatory and 6π disrotatory electrocyclic cascade under the Pd(II)-catalyzed isomerization conditions. Then a retro [2 + 2] cycloaddition^{5c} of **1** generated **2** (Scheme 1).

Recently, we demonstrated that the addition of either a Zn(II)-type or NAD^+ -dependent HDAC inhibitor in the cultivation of fungi, especially 500 μM of suberoyl bis-hydroxamic acid (SBHA) (a Zn(II) type HDAC inhibitor) and 100 μM of nicotinamide (an NAD^+ -dependent HDAC inhibitor), significantly enhanced polyketide production, and we successfully isolated a variety of novel natural products.¹⁰ Thus, we applied these chemicals to the *Penicillium* cultivation. **1** and **3** in the mycelium were notably enriched upon cultivation with SBHA (500 μM) or nicotinamide (100 μM) (Table 2). Upon production of **4**, nicotinamide showed its producing effect, while SBHA did not show a distinct effect.

Scheme 1. Plausible Biogenetic Pathway of **1**, **2**, and **4** from **3**

(10) (a) Asai, T.; Yamamoto, T.; Oshima, Y. *Tetrahedron Lett.* **2011**, 52, 7042–7045. (b) Asai, T.; Dan, L.; Obara, Y.; Taniguchi, T.; Monde, K.; Yamashita, K.; Oshima, Y. *Tetrahedron Lett.* **2012**, 53, 2239–2243. (c) Asai, T.; Chung, Y. M.; Sakurai, H.; Ozeki, T.; Chang, F. R.; Yamashita, K.; Oshima, Y. *Org. Lett.* **2012**, 54, 513–515. (d) Asai, T.; Yamamoto, T.; Oshima, Y. *Org. Lett.* **2012**, 54, 2006–2009. (e) Asai, T.; Morita, S.; Shirata, N.; Taniguchi, T.; Monde, K.; Sakurai, H.; Ozeki, T.; Oshima, Y. *Org. Lett.* **2012**, 54, 5456–5459.

Table 2. Effects of HDAC Inhibitors on the Concentrations of **1**, **3**, and **4** in the Mycelium (Dry Weight)^a

condition	1 (mg/g)	3 (mg/g)	4 (mg/g)	mycelium ^b (g)
control	0.13	0.24	7.82	0.39
SBHA (500 μ M)	0.72	1.31	8.22	0.36
nicotinamide (100 μ M)	1.32	3.07	20.62	0.24

^a Each compound was quantified against external standards. Quantification was based on peak area in the reversed-HPLC chromatogram.

^b In each condition, the fungus was cultivated in 60 mL of culture medium. Then the mycelia were collected and dried.

The cell growth inhibitory activities of **1–4** were evaluated in human HCT 116 cells. Citreoviripyrone A (**1**) exhibited a moderate inhibitory activity on cell growth with a GI_{50} value of 10.4 μ M using the luciferin/luciferase assay (Cell Titer Glo, Promega) to quantitate the ATP of live cells. However, **1** did not inhibit growth within 50 μ M in an MTT assay.

In conclusion, we isolated a novel α -pyrone polyketide with bicyclo[4.2.0]octadiene, citreoviripyrone A (**1**), from a (–)-citreoviridin (**4**)-producing endophytic *Penicillium* fungus. Using Pd(II) catalyzed polyene isomerization of citreomontanin (**3**) realized the biomimetic formation of **1** and **2** via an uncommon natural product biosynthetic pathway. We also found that both Zn(II)-type and NAD^+ -dependent HDAC inhibitors could activate these α -pyrone polyketide productions in the endophytic fungus and verified the availability of HDAC inhibitors in the natural compound production.

Acknowledgment. This work was supported in part by a Grant-in-Aid for Scientific Research (Nos. 23710248 and 23590582) from the Ministry of Education, Science, Sports and Culture of Japan; A-STEP (FS-stage) (No. AS231Z01347G) from Japan Science and Technology Agency (JST); the Uehara Memorial Foundation; and Japan Association for Chemical Innovation.

Supporting Information Available. Experimental methods, and NMR spectra of new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.